



A Course of Analytical Chemistry, by W.W. Pink and G.E. Webster (Paperback)

By William W Pink

Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1874 edition. Excerpt: . Acid. 121. Add MnO₂, manganic oxide (black oxide of manganese), and SO₂, Ho₂, then apply heat; evolution of chlorine takes place, known by its smell, colour, c. 122. NO₂Ag_o gives a curdy-looking white ppt. of AgCl, argentic chloride (chloride of silver), soluble in AmHo, but insoluble in weak NO₂Ho; exposure to the sunlight always darkens this ppt. HBr.-- Hydrobromic Acid. 123. Add MnO₂ and SO₂Ho₂, then apply heat; evolution of bromine vapour takes place, which colours moistened starch paper ytlhw. 124. NO₂Ag_o gives a cream-coloured yellow ppt. of AgBr, sightly soluble in AmHo, but insoluble in N₀₂Ho. HI.--Hydriodic Acid. 125. Treat as in Reaction No. 123; evolution of iodine vapour of a violet or purple colour takes place, which colour moistened starch paper blue. 126. NO₂Ag_o gives a light yellow ppt. of AgI, argentic iodide (iodide of silver), soluble in KCy, but insoluble in...



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